



# Citrate, Calcium and Magnesium metabolism and related toxicities

**Hans Vrielink, MD, PhD**

Sanquin Blood Supply, Amsterdam, The Netherlands

European Society for Hemapheresis

Joint Task Force for Apheresis Education and Certification

**Marleen Neyrinck, RN**

AZ Delta, Roeselare, Belgium

European Society for Hemapheresis

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Blood and Beyond



## Disclosure of Relevant Financial Relationships

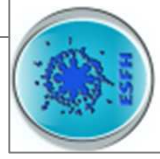
- Hans None
- Marleen None





## Adverse reactions

(Related to donor and patient apheresis)



## Applied Physiology in Apheresis

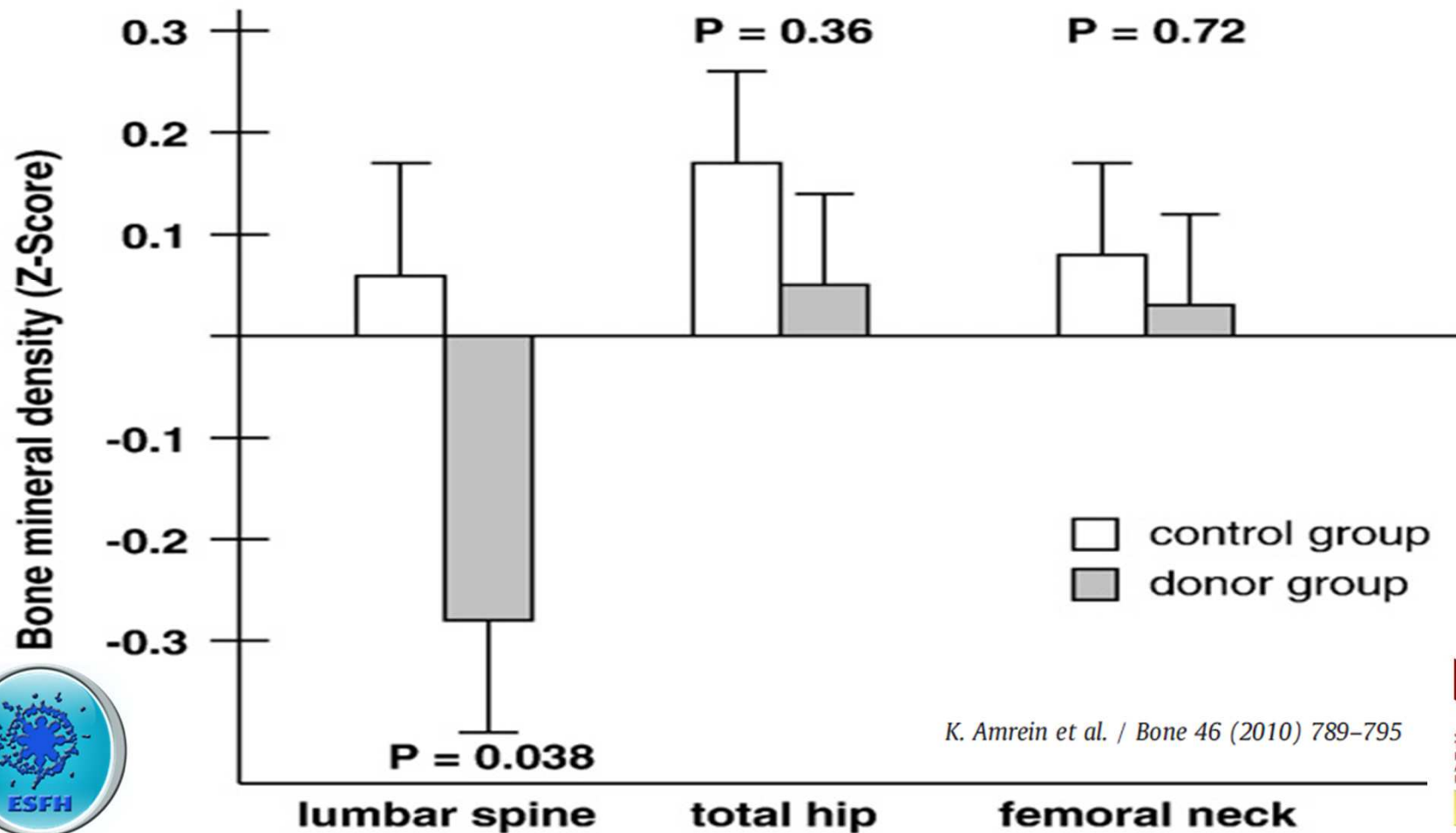


## Long term effects of apheresis procedures

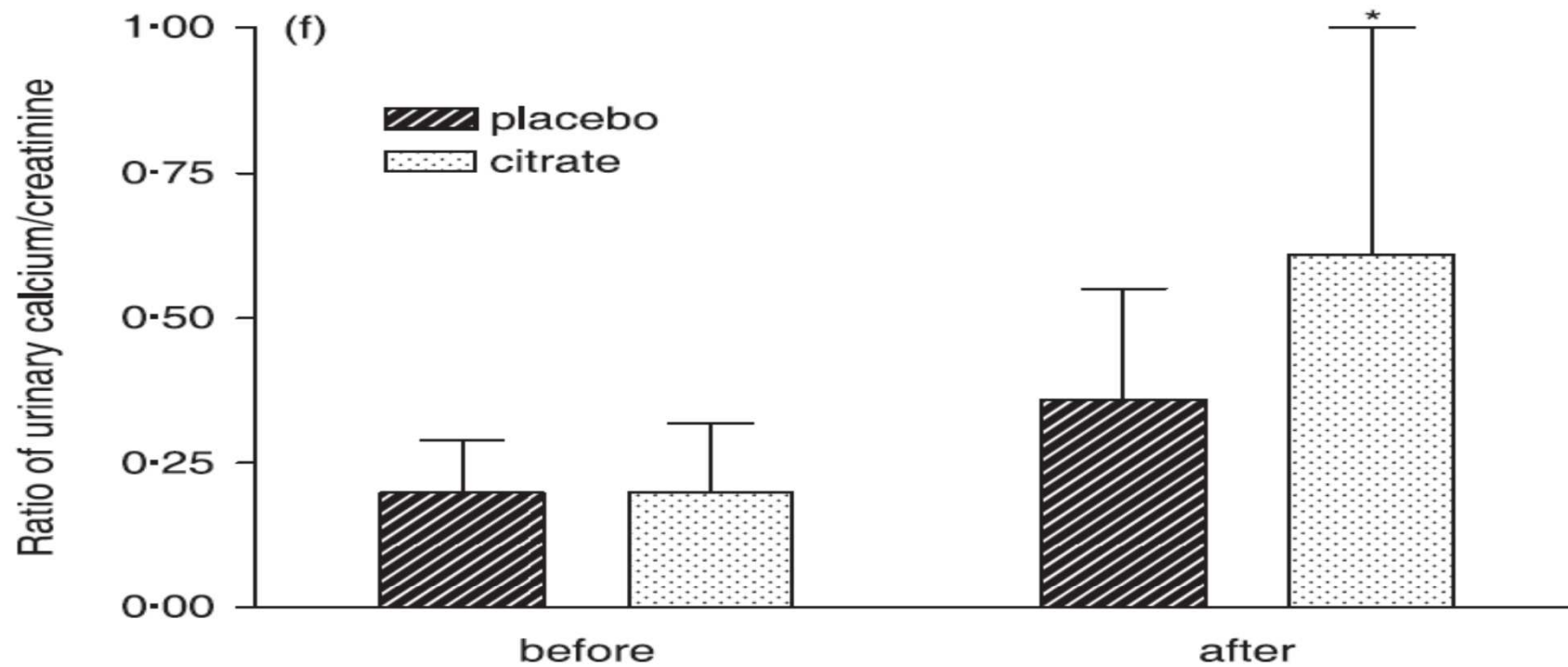
- Comparison bone density of 45 donors >100 PLT-apheresis with 40 donors <50 procedures.
- 35% of >100 procedures donors showed significant osteoporosis.

Dettke J. Clin Apheresis 2003

## Long term effects in frequent plt donors?



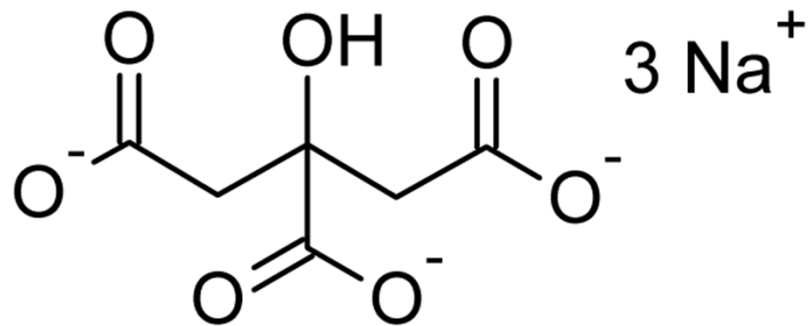
## Calcium in urine



Comparison of urinary calcium excretion  
Excretion of urinary calcium increase after citrate  
intervention. \* $P = 0.034$

Y. Chen *et al.* *Vox Sanguinis* (2009) 97, 324–329

## Tri-sodiumcitrate and calcium



[Vhttp://4.bp.blogspot.com/-69-2fUgntos/T9UE0fnWxTI/AAAAAAAAAMPQ/XwbSSOWjNZs/s1600/Calcium.jpg](http://4.bp.blogspot.com/-69-2fUgntos/T9UE0fnWxTI/AAAAAAAAAMPQ/XwbSSOWjNZs/s1600/Calcium.jpg)

## Calcium

- 99% in the bones → calcium phosphate  
(± 24,500 mmol)



Extra cellular fluid: 22.5 mmol → 9 mmol in plasma (2.2 – 2.6 mmol/L)



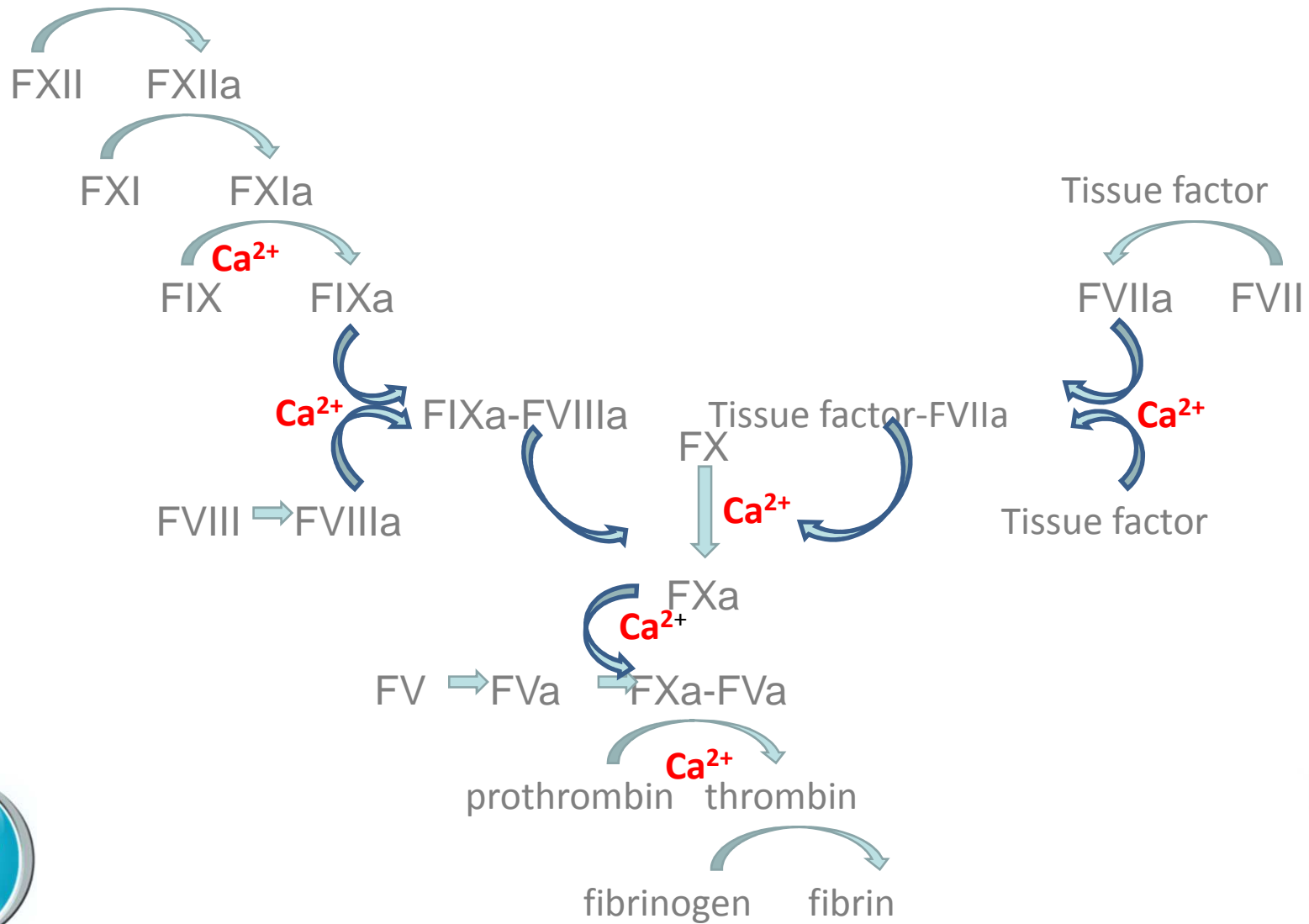
## Plasma calcium

Total calcium 2.2 – 2.6 mmol/L (9 - 10.5 mg/dL)

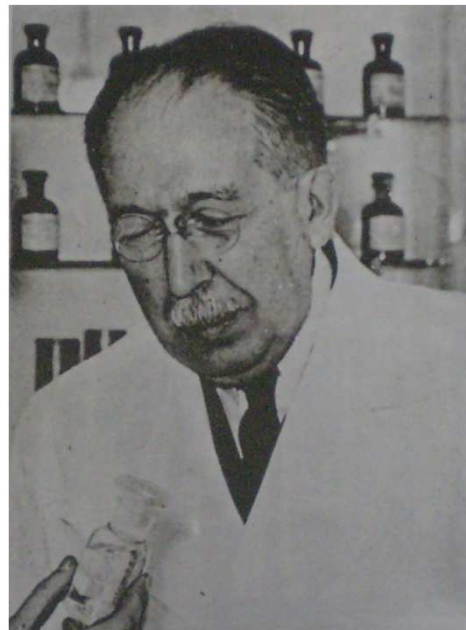
- Ionized (free) calcium 1.1 – 1.4 mmol/L (4.5 – 5.6 mg/dL)
- Remainder bound mainly to albumen ( $\pm$  50%)

## Function of Calcium

- Structural function → bones
- Signaling function → messenger for some hormones
- Enzymatic function → co-enzyme for clotting factors
- Function in transmission of nerve impulse
- Function in the contraction of muscles
- Stabilization of cellular membranes

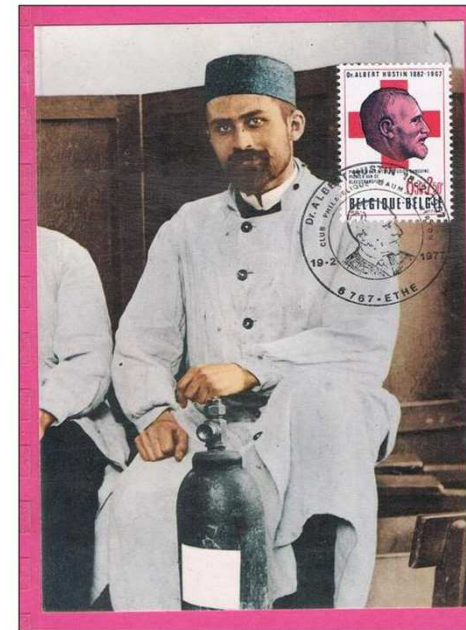


## Prevention of blood clotting by citrate

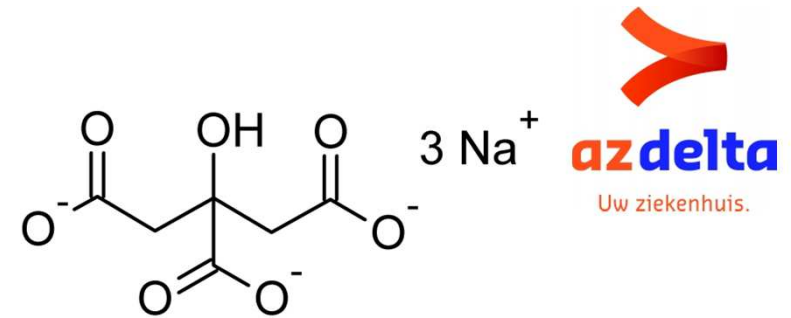


Luis Agote

1914



Albert Hustin



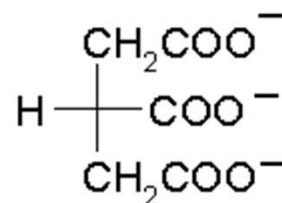
## Citrate

- Trisodiumcitrate
  - Prevention of blood clotting in disposable / machine
  - Flavoring and buffering agent in drinks / food (E330)
  - Laxative
  - WHO “oral rehydration solution”

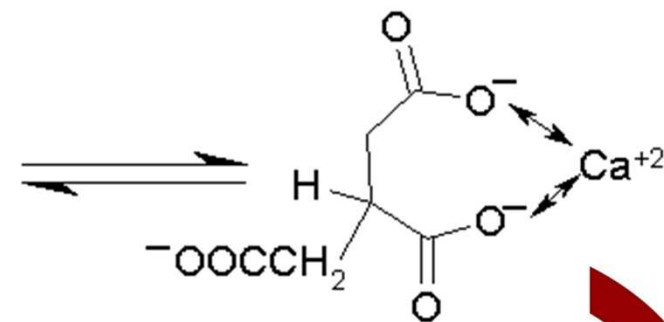


## Citrate handling during apheresis procedures

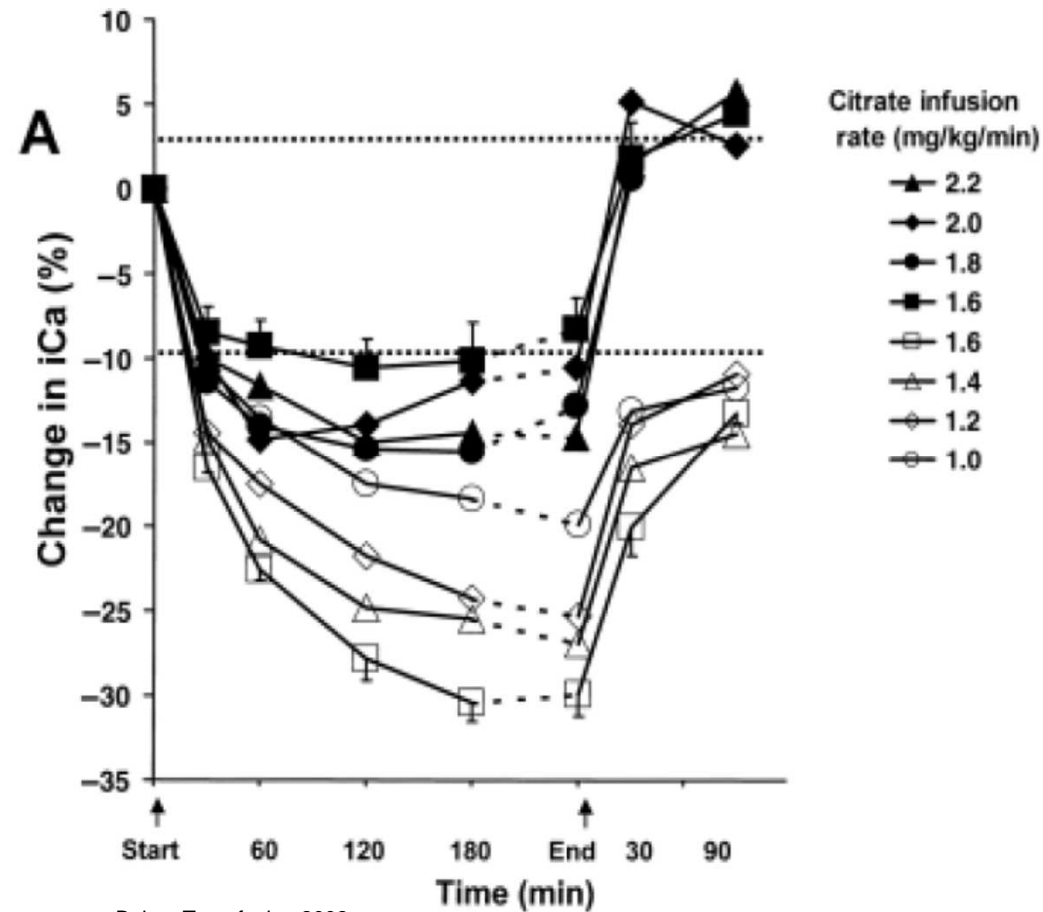
- Tri-sodiumcitrate is added to whole blood in procedure specific ratio
- Citrate resolves completely in plasma
- Citrate chelates free Calcium and Magnesium
- Citrate returns to donor with plasma containing components



sodium citrate

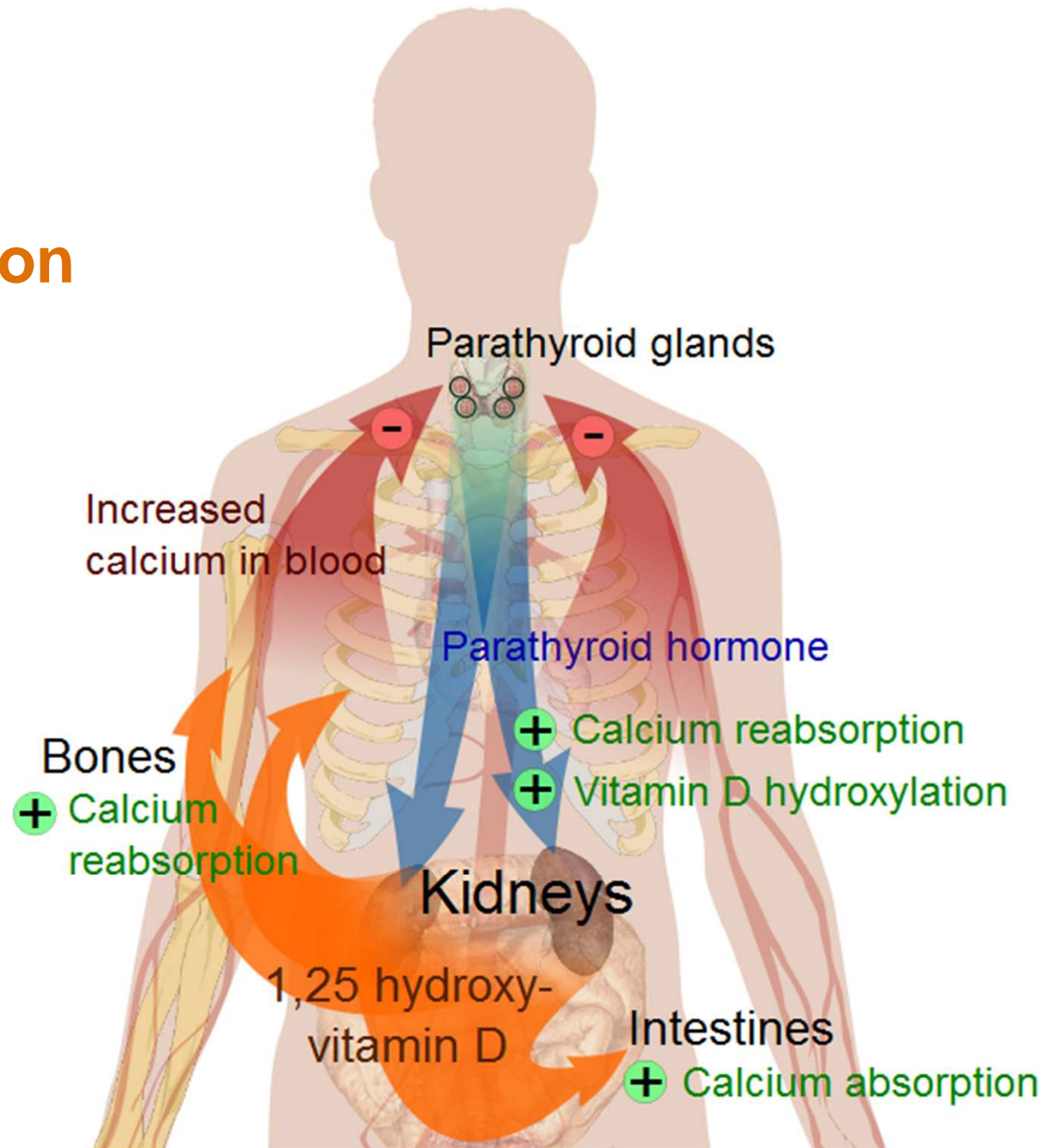


# Serum calcium & citrate infusion



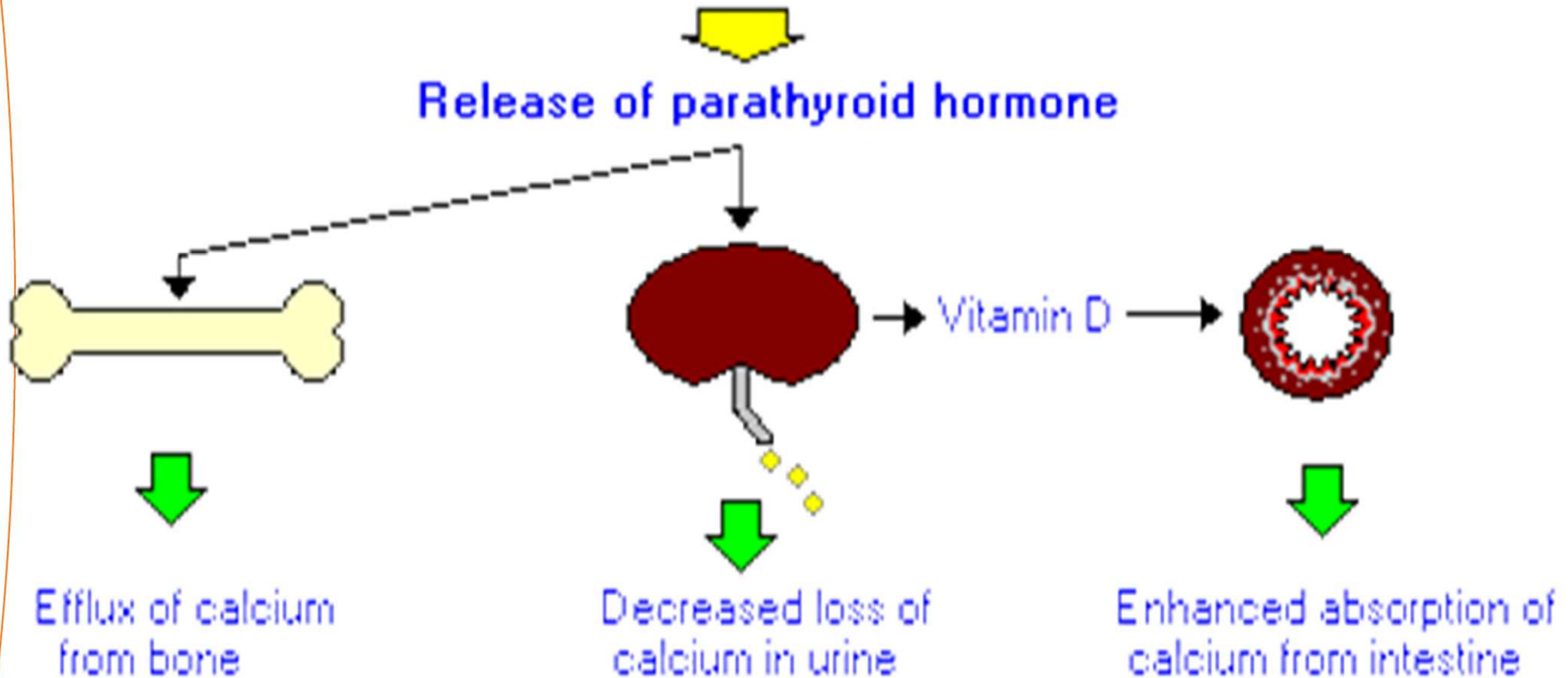
Bolan, Transfusion 2002

# Calcium regulation



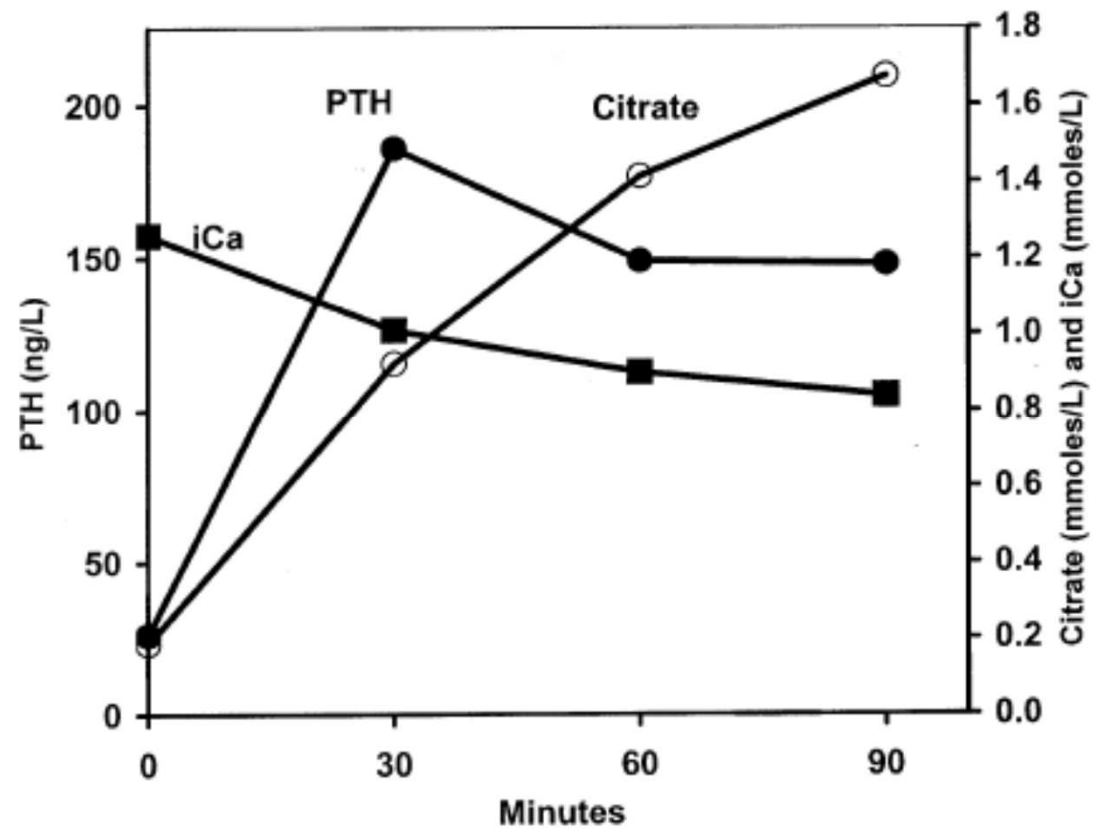


## Low concentration of calcium in blood



## Increased concentration of calcium in blood

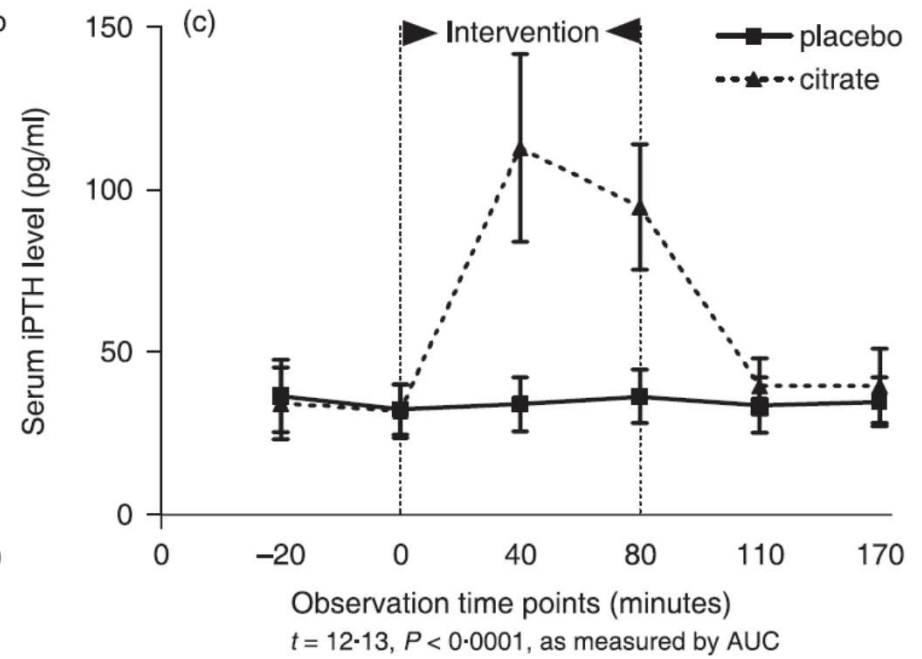
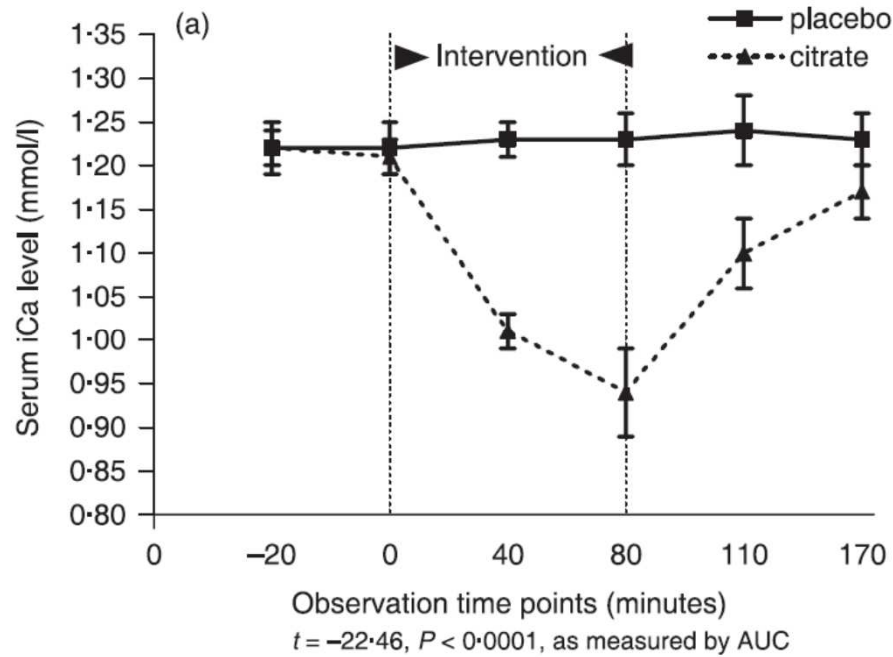
## Serum PTH, iCa and Citrate during plt apheresis



McLeod BC, Szczepiorkowski ZM, Weinstein R, Winters JL, eds; *Apheresis: Principles and Practice*, 3rd edition; Bethesda, MD: AABB Press, 2010



# Serum PTH and Calcium



Y. Chen *et al.* *Vox Sanguinis* (2009) 97, 324–32!

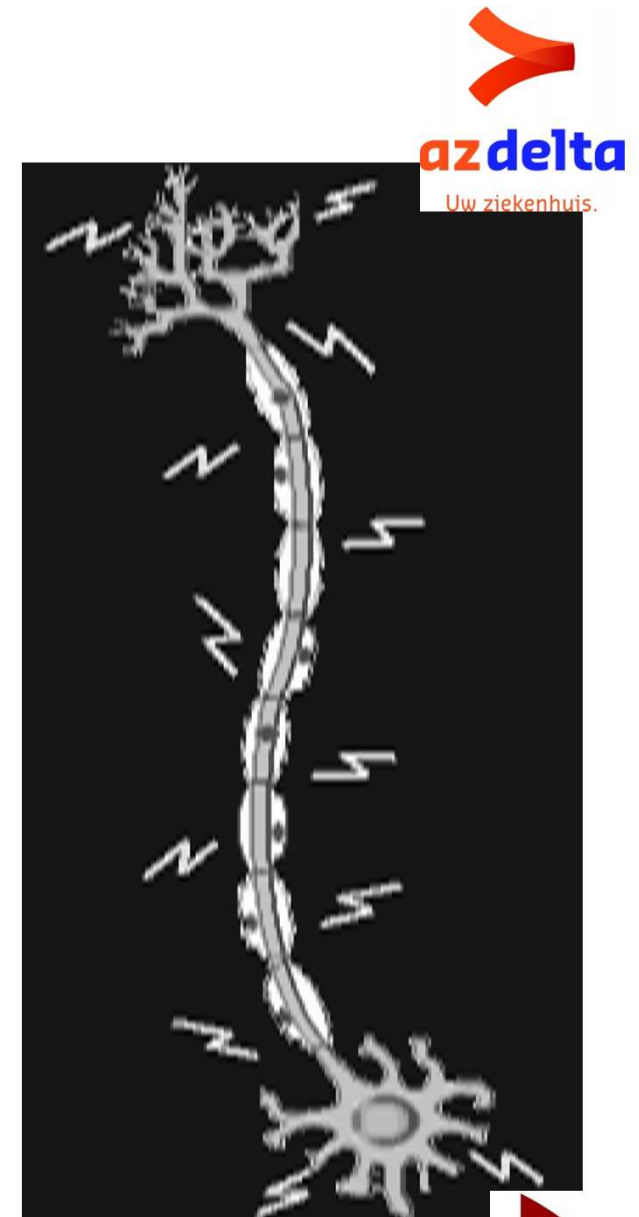


## Function of Calcium

- Structural function → bones
- Signaling function → messenger for some hormones
- Enzymatic function → co-enzyme for clotting factors
- Function in transmission of nerve impulse
- Function in the contraction of muscles

## Hypocalcemia

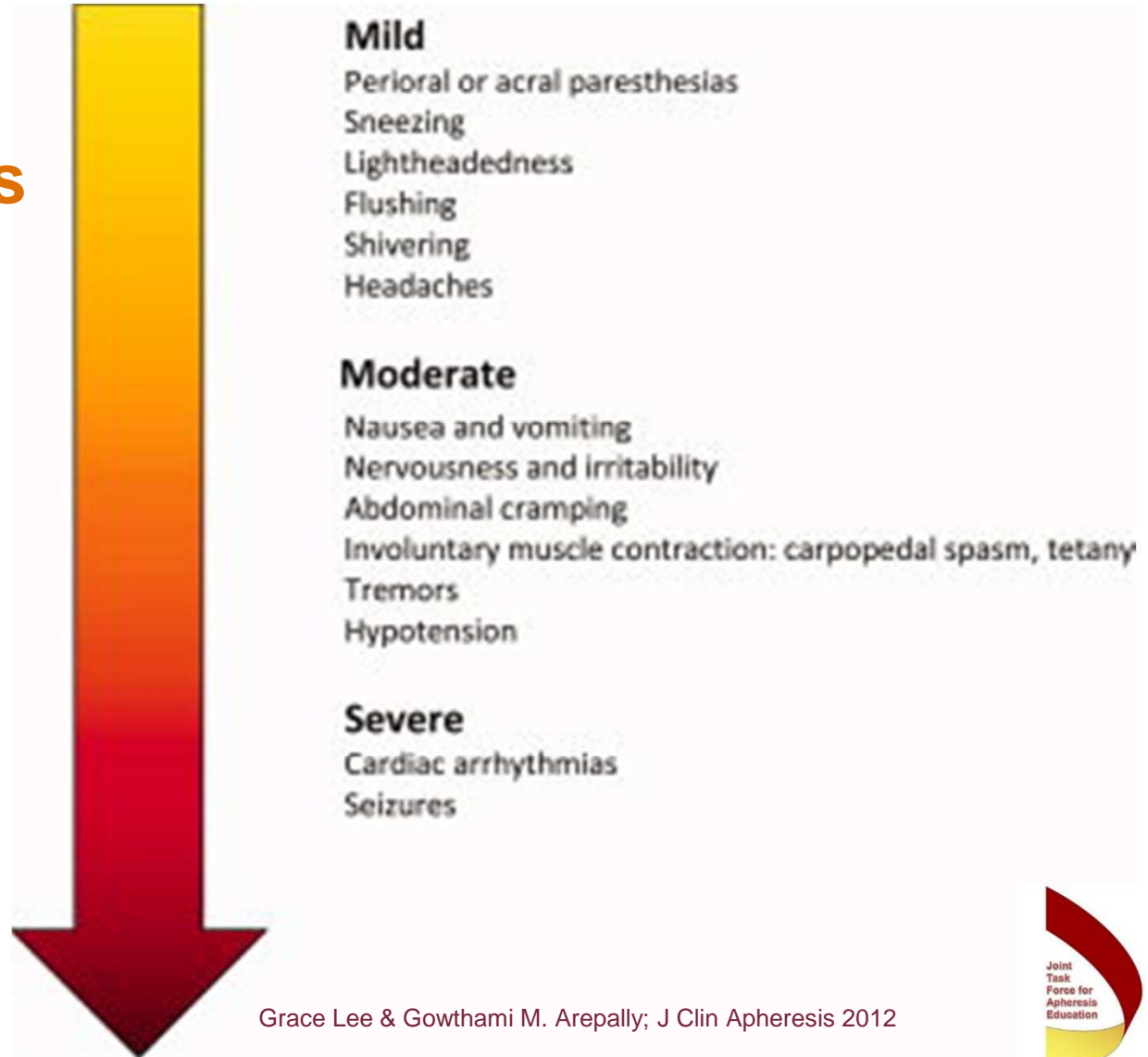
- Decrease in ionized calcium results in increased excitability of neurons to the point of spontaneous depolarization.



## Function of Calcium

- Structural function → bones
- Signaling function → messenger for some hormones
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# Symptoms of citrate reactions



## Citrate

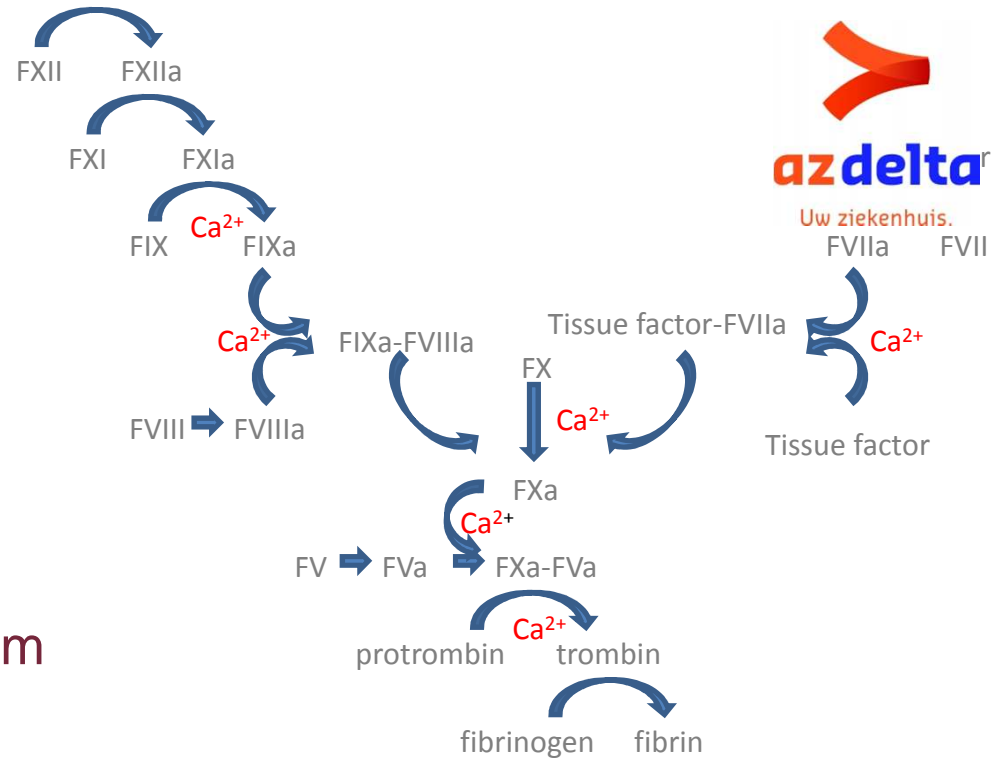
- Completely dissolved in plasma
- No binding to cells
- Chelates calcium and Magnesium
- Prevents coagulation

Neutralized by:

- Distribution throughout extra cellular fluid
- Excretion by the kidneys
- Rapid metabolism by the kidneys, liver and skeletal muscle



Jeff Winters, AABB Oct. 2007

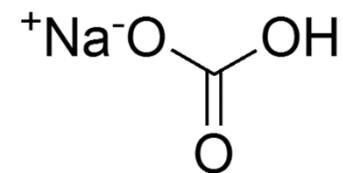
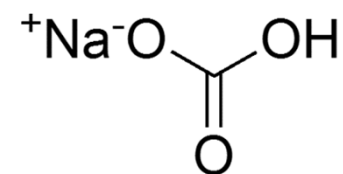
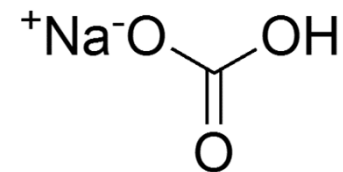
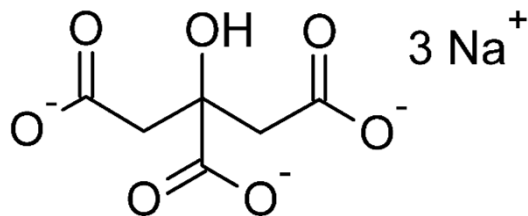




## Citrate → Metabolic Alkalosis

Additional factors to consider

- Presence of citrate in replacement fluids (e.g. FFP for TTP)
- Large volume PBSC collections (average drop  $\text{Ca}^{2+}$  of  $11.3 \pm 7\%$ )
- Citrate → Bi-carbonate → Alkalosis



$\text{Ca}^{2+}$

$\text{Mg}^{2+}$



## Citrate → Metabolic Alkalosis & Hypokalemia

Additional factors to consider

- Presence of citrate in replacement fluids (e.g. FFP for TTP)
- Large volume PBSC collections (average drop  $\text{Ca}^{2+}$  of  $11.3 \pm 7\%$ )
- Citrate → Bi-carbonate → Alkalosis → excretion citrate ↑
- Renal disease preventing the excretion of bicarbonate and citrate  
→ symptoms of hypocalcemia ↑, suppression of respiratory rate  
→ **Metabolic alkalosis**
- Metabolic alkalosis results in potassium uptake in cells → hypokalemia  
Decrease in potassium and cardiac arrhythmia

## Citrate

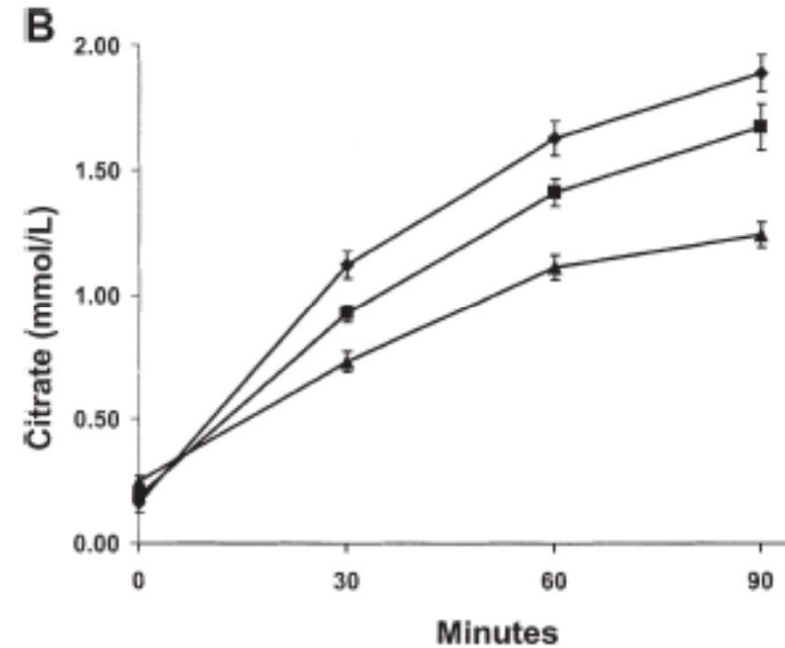
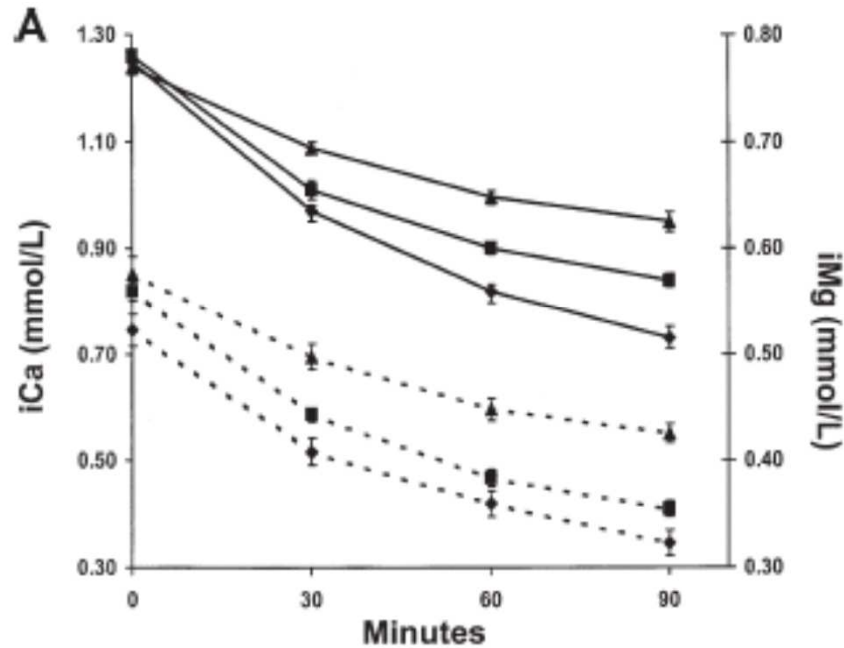
- Chelates calcium and Magnesium



## Citrate → Hypomagnesemia

- $Mg^{2+}$  also bound by citrate
- During plateletapheresis: 30% drop in magnesium levels

# Serum magnesium during plateletapheresis



Bolan, et al. Transfusion 2001;41:1165-1171

## Plasma magnesium

Total magnesium 0.7 – 1.1 mmol/L (1.5 – 2.5 mg/dL)

- Ionized (free) magnesium 0.5 – 0.7 mmol/L (1.1 – 1.5 mg/dL)
- Remainder bound mainly to Albumen and globulins

# Magnesium

Involved in:

- Synthesis of nucleic acids
- Synthesis of proteins
- Intermediary metabolism

- Specific actions in
  - Neuromuscular systems
  - Cardiovascular systems

$Mg^{2+}$  competes with  $Ca^{2+}$  for binding sites on proteins and membranes

Competitively inhibition of calcium





# Magnesium

Affects:

- Muscular contraction and relaxation including the heart and vascular muscles.
- Electrical activity of myocardial cells
- Stabilization of the axon
- The release of neurotransmitters

## Hypomagnesaemia

Caused by

- Redistribution of Mg
  - Gastrointestinal
  - Renal loss
  - Renal disease
  - Endocrinal
  - Diabetes mellitus
  - Alcoholism
  - Miscellaneous
- Drugs
    - Diuretics
    - Cytotoxic drugs
    - Antibiotics
    - B adrenergic agents
    - Others

# Hypomagnesaemia

Citrate and hypocalcaemia

## Electrolyte disturbance

Hypokalaemia  
Hypocalcaemia

## Neuromuscular and central nervous system

Carpopedal spasm  
Convulsions  
Muscle cramps  
Muscle weakness, fasciculations, tremors  
Vertigo  
Nystagmus  
Depression, psychosis  
Athetoid movements & choreiform movements

## Cardiovascular

Atrial tachycardias, fibrillation  
Supraventricular arrhythmias  
Ventricular arrhythmias  
Torsade de pointes  
Digoxin sensitivity

## Complications of magnesium deficiency

Altered glucose homeostasis  
Atherosclerotic vascular disease  
Hypertension  
Myocardial infarction  
Osteoporosis

## Miscellaneous

Migraine  
Asthma  
Chronic fatigue syndrome  
Impaired athletic performance

## Citrate → Hypomagnesemia

- $Mg^{2+}$  also bound by citrate
- During plateletapheresis: 30% drop in magnesium levels
- Steeper decrease and recovers more slowly than calcium
- Muscle spasms & weakness
- Decreased vascular tonus (blood pressure) + abnormal cardiac contractility
- Interference with potassium and calcium homeostasis
- If suspected 8 mmol  $Mg^{2+}$  i.v. in 1 minute

## In Summary

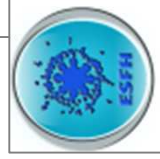
- For apheresis procedures we use citrate solutions
- Citrate is binding calcium and magnesium
- Citrate is returned to the donor
- Hypocalcemia and hypomagnesemia → effects
- Direct (side) effects of citrate





## Adverse reactions

(Related to donor and patient apheresis)



## Applied Physiology in Apheresis



## For information



<http://www.sanquin.nl/en/products-services/consulting-services>



[h.vrielink@sanquin.nl](mailto:h.vrielink@sanquin.nl)